Compounds, Elements, and Molecules

S1-2-09 Compare elements to compounds. Include: atoms, molecules

We must be able to differentiate between the following terms:

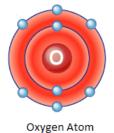
Atom:

- Smallest **UNIT** of an **ELEMENT**.
- Made of **PROTONS**, **ELECTRONS** and **NEUTRONS**

Examples:







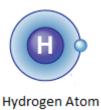


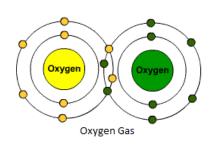


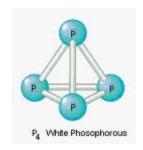
Element

- A **PURE SUBSTANCE** made of **IDENTICAL ATOMS**.
- Cannot be <u>BROKEN DOWN</u> into <u>DIFFERENT KINDS</u> of atoms.
- Elements are made of atoms

Examples:





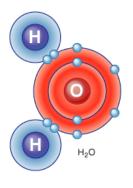


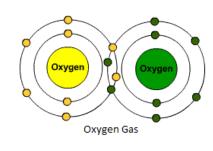
Molecule

- Is a <u>PURE SUBSTANCE</u> made of a <u>CLUSTER</u> of atoms of <u>SIMILAR</u> or <u>DIFFERENT ELEMENTS</u>.

- Can be **BROKEN DOWN** into those **ATOMS** during a **CHEMICAL CHANGE**.

Examples:



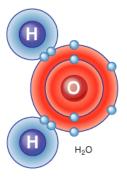


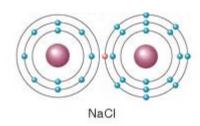
Compound

- A <u>PURE SUBSTACNCE</u> whose <u>MOLECULES</u> are made of <u>DIFFERENT KINDS</u> of <u>ATOMS</u>.

COMPOUNDS can be broken down into <u>DIFFERENT ELEMENTS</u>.

Examples:





Recall the Particle Theory of Matter:

- A particle is a small "THING".
- ALL MATTER is composed of tiny PARTICLES called ATOMS.
- **ATOMS** can join to form **MOLECULES** or **COMPOUNDS** which are also **PARTICLES**.

A **PURE SUBSTANCE** is something that contains only **ONE KIND** of **MATTER**.

A **MIXTURE** is something that contains **TWO** or **MORE PURE SUBSTANCES**.

There are about 10 million known **PURE SUBSTANCES**

- Only about <u>112</u> of these are actually <u>ELEMENTS</u>
- The rest are **COMPOUNDS** or **MOLECULES**

Examples

Baking soda is actually **SODIUM BICARBONATE** (NaHCO₃)

- → A mixture of **SODIUM**, **CARBON** and **OXYGEN** atoms.
- → It is a **COMPOUND** or a **MOLECULE**

Pure OXYGEN is just O_2 ,

→ It is a **MOLECULE** AND an **ELEMENT**.

Salt is also known as **SODIUM CHLORIDE** (NaCl)

→ A **COMPOUND** or **MOLECULE**

Toothpaste is also known as **SODIUM FLUORIDE** (NaF)

→ A <u>COMPOUND</u>/<u>MOLECULE</u>

What is the composition of other household items?

Vinegar – ACETIC ACID $(HC_2H_3O_2)$

Drain Cleaner – **SODIUM HYDROXIDE** (**NaOH**)

Chalk – **CALCIUM CARBONATE CaCO**₃

Lime – CALCIUM OXIDE CaO

Ammonia – NH₃

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Can you think of any pure substances that you see every day?

Examples:

Put the above substances in the correct column:

Element

Compound