## Compounds, Elements, and Molecules

We must be able to differentiate between the following terms:

## Atom:

- Smallest UNIT of an ELEMENT.
- Made of PROTONS, ELECTRONS and NEUTRONS

Examples:


## Element

- A PURE SUBSTANCE made of IDENTICAL ATOMS.
- Cannot be BROKEN DOWN into DIFFERENT KINDS of atoms.
- Elements are made of atoms

Examples:


## Molecule

- Is a PURE SUBSTANCE made of a CLUSTER of atoms of SIMILAR or DIFFERENT ELEMENTS.
- Can be BROKEN DOWN into those ATOMS during a CHEMICAL CHANGE.

Examples:


## Compound

- A PURE SUBSTACNCE whose MOLECULES are made of DIFFERENT KINDS of ATOMS.
- COMPOUNDS can be broken down into DIFFERENT ELEMENTS.

Examples:


## Recall the Particle Theory of Matter:

- A particle is a small "IHING".
- ALL MATTER is composed of tiny PARTICLES called ATOMS.
- ATOMS can join to form MOLECULES or COMPOUNDS which are also PARTICLES.

A PURE SUBSTANCE is something that contains only ONE KIND of MATTER.
A MIXTURE is something that contains TWO or MORE PURE SUBSTANCES.
There are about 10 million known PURE SUBSTANCES

- Only about 112 of these are actually ELEMENTS
- The rest are COMPOUNDS or MOLECULES


## Examples

Baking soda is actually SODIUM BICARBONATE ( $\mathrm{NaHCO}_{3}$ )
$\rightarrow$ A mixture of SODIUM, CARBON and OXYGEN atoms.
$\rightarrow$ It is a COMPOUND or a MOLECULE

Pure OXYGEN is just $\mathbf{O}_{2}$,
$\rightarrow$ It is a MOLECULE AND an ELEMENT.
Salt is also known as SODIUM CHLORIDE ( NaCl )
$\rightarrow$ A COMPOUND or MOLECULE
Toothpaste is also known as SODIUM FLUORIDE (NaF)
$\rightarrow$ A COMPOUND/MOLECULE

What is the composition of other household items?

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Vinegar - ACETIC ACID (HC2 H
Drain Cleaner - SODIUM HYDROXIDE (NaOH)
Chalk - CALCIUM CARBONATE CaCO
Lime - CALCIUM OXIDE CaO
Ammonia - NH3
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## Pure Substances...

Can you think of any pure substances that you see every day?

## Examples:

Put the above substances in the correct column:

