

#### **Outcome:**

- Outline the transformation of alkenes to alkynes and vice versa.
- Name, Draw and construct molecular models of alkynes and branched alkynes.

## <u>lkynes:</u>

#### **Alkynes:**

Aliphantic Compounds formed from CHAINS of carbon atoms that have one or more **TRIPLE BONDS**.

Ex) **Propyne** 

I - Butyne

- C= C- C-- C = (- - - - -(344 Cy HG

- Notice that there is a relationship between the number of carbons and hydrogens...
  - They have the general formula

 $\underline{C}_{n}\underline{H}_{2n-2}$ 

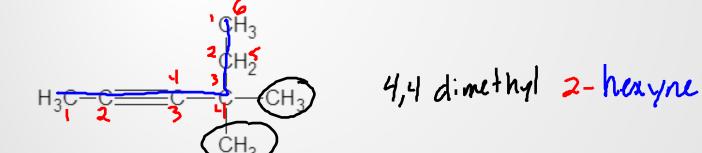
- Also called UNSATURATED HYDROCARBONS (TRIPLE BONDS).
- Are also ALIPHATIC HYDROCARBONS.

## **Naming Alkynes:**

Same rules as for Alkenes, but end in the suffix "YNE".

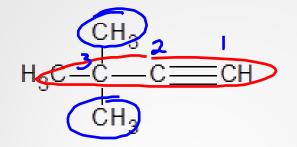
**Examples:** 

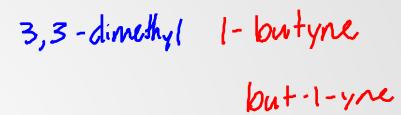


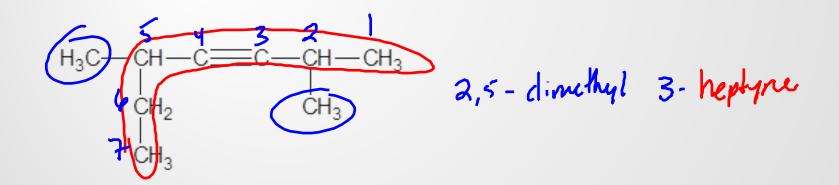


### **Naming Alkynes:**

Try these...





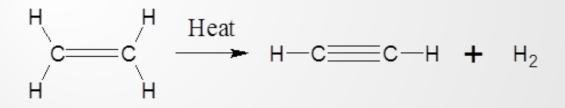


# **Reactions of Alkynes:**

Alkanes can also undergo chemical reactions like the alkanes and alkenes...

#### **Dehydrogenation**

• We can dehydrogenate <u>ALKENES</u> to make <u>ALKYNES</u>:



#### **Reactions of Alkynes:**

#### **Hydrogenation**

• We can hydrogenate **<u>ALKYNES</u>** to make **<u>ALKENES</u>**:

$$H - C \equiv C - H + H_2 \xrightarrow{\text{Ni, Pt, Pd}} H = C = C + H_2$$

We can also hydrogenate <u>ALKYNES</u> to make <u>ALKANES</u>

$$H - C \equiv C - H + 2H_2 \xrightarrow{Ni, Pt, Pd} H - C - C - H$$